

MAHATMA GANDHI COLLEGE

THIRUVANANTHAPURAM

POST GRADUATE DEPARTMENT OF CHEMISTRY & RESEARCH CENTER

NOTICE

ADD-ON COURSE -MOLECULAR SPECTROSCOPY

Department of chemistry is conducting an Add-On course titled "MOLECULAR SPECTROSCOPY" for the academic year 2018-2019. This course aims to provide students with an in-depth understanding of spectroscopic techniques and their practical applications.

Course Duration: 30 hour

Last Date for Registration: 4/06/18

For any queries or additional information, please contact the course coordinator, Head, Department of Chemistry.

Head, Department of Chemistry

P.G. Dept. of Chemistry & Research Cent. Mahatma Gandhi College
Thiruvananthapuram

Add on course Syllabus

Molecular spectroscopy

Unit 1 Introduction

8h

General features of absorption - Beer-Lambert's law and its limitations, transmittance, Absorbance, and molar absorptivity. Single and double beam spectrophotometers. Electronic spectroscopy: Interaction of electromagnetic radiation with molecules and types of molecular spectra. Energy levels of molecular orbitals (σ, π, n) . Selection rules for electronic spectra. Types of electronic transitions in molecules effect of conjugation. Concept of chromophore and auxochrome.

UNIT-II Infra red spectroscopy

8h

Different Regions in Infrared radiations. Modes of vibrations in diatomic and polyatomic molecules. Characteristic absorption bands of various functional groups. Interpretation of spectra-Alkanes, Aromatic, Alcohols, carbonyls, and amines with one example to each.

UNIT-III Proton magnetic resonance spectroscopy

8h

Principles of nuclear magnetic resonance, equivalent and non-equivalent protons, position of signals. Chemical shift, NMR splitting of signals - spin-spin coupling, coupling constants. Applications of NMR with suitable examples - ethyl bromide, ethanol, acetaldehyde, 1,1,2-tribromo ethane, ethyl acetate, toluene and acetophenone.

UNIT-IV

6h

Instrument demonstration

UV-Visible spectroscopy, IR spectroscopy, Photoluminescence Spectrometer (PL)

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Course Outcome

- Predict which organic compounds should exhibit visible color based upon extent of conjugation.
- Explain the origin of infrared absorptions in terms of vibrational modes of covalent bonds.
- Predict direction of chemical shifts caused by various structural shielding or deshielding effects.
- Explain the principle and instrumentation of electronic spectroscopy, Infrared spectroscopy, NMR spectroscopy and analyze the spectra of different species.

| Physic Deplement on Assessment | Academic Year 2018-2 | |
|--------------------------------|--|--|
| SI.No. | Name | Candidate Code |
| 1 | Abhirag G.R | 235118506234 |
| 2 | Amal Krishnan | 235118524275 |
| 3 | Amritha 8.5 | 235118526686 |
| d management | Amritha M.B | 235118520216 |
| 5 | Anjana A.V | 235118574223 |
| 6 | Anjana M.G | 235118516210 |
| 7 | Aparna S | 235118528245 |
| 8 | Bharath S Kumar | 235118566031 |
| 9 | Jithin Aji | 235118525399 |
| 10 | Kavya M Biju | 235118564973 |
| 11 | Keerthana R.S | 235118510400 |
| 12 | Maneesha M | 235118523632 |
| 13 | Midhun Raj I | 235118537215 |
| 14 | Neetha J.A | 235118570529 |
| 15 | Parvathy J | 235118519510 |
| 16 | Pradeep Sankar | 235118500234 |
| 17 | Sruthy M.R | 235118550516 |
| 18 | Surya S.S | 235118578946 |
| 19 | Yadukrishnan R.S | 235118501126 |
| 20 | Archa Santhosh | 235118560338 |
| 21 | Adithya A | 235118536743 |
| 22 | Akhil S | 235118509782 |
| 23 | Aneesh S.R | 235118538714 |
| 24 | Anoop M | 235118591403 |
| 25 | Arunima Chandra | 235118511593 |
| 26 | Athira Anil | 235118520867 |
| 27 | Badhush M.S | 235118526247 |
| 28 | Devika Venu J | 235118548074 |
| 29 | Govind G.S | 235118551705 |
| 30 | Greeshma G | 235118508603 |
| 31 | Jithin M.V | 235118800292 |
| 32 | Kallyani M | 235118500292 |
| 33 | Lekshmy A.K | |
| 34 | | 235118575757 |
| | Lekshmy S.S | 235118574076 |
| 35 | Megha M Nair | 235118572376 |
| 36 | Neethu Lekshmy R.J | 235118548320 |
| 37 | Nikhil Rajendran | 235118527693 |
| 38 | Rajikrishna R.C | 235118561819 |
| 39 | Reshma S Nair | 235118556305 |
| 10 | Sreelekshmy R | 235118507775 |
| 41 | Sruthy S | 235118524624 |
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| 14 | Vrinda V.S | 235118508098 |
| 45 | Amritha R.V | 235118538348 |
| 46 | Anusha Raj | 235118536117 |
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| 47 | Aravind M Raj | 235118546153 |



| 49 | Harikrishnan V | 235118526060 |
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| 50 | Krishnanunni V.S | 235118522296 |
| 51 | Krishnapriya R | 235118501405 |
| 52 | J.D Haripriya | 235118559736 |
| 53 | Aiswarya Krishna | 235118524092 |
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Add on Course Examination Molecular Spectroscopy MCQ Answer all the questions

Marks: 10

Time: 15 minutes

 $\mathbb{C}_{a}(4) < (3) < (2) < (1)$

5.

| Absorption of radiation in the UV range attributable to n→π* electronic transitions is |
|--|
| 1. Absorption of radiation in the CV range arrivated of characteristic of which of the following types of compounds? |
| C a) Aromatic hydrocarbons. |
| Unsaturated carbonyl compounds. |
| C Non-conjugated polyenes. |
| ES |
| Conjugated polyenes. Which is the correct order of increasing wave number of the stretching vibrations of (1) C-H (alkane), (2) C-H (alkene), (3) C-H (alkyne), and (4) C-H (arene)? |
| $\mathbb{C}_{-3}(1) < (2) \approx (3) < (4)$ |
| $\mathbb{C}_{-\{1\}}(4) < (3) \approx (2) < (1)$ |
| $\mathbb{C}_{(3)}(3) < (4) \approx (2) < (1)$ |
| $\mathbb{C}_{-(1)}(1) < (4) \approx (2) < (3)$ 3. Which of the following statements in the context of 'H NMR spectroscopy is true? |
| C a) Arene C-H chemical shift (δ) values are greater than simple alkenes C-H chemical shift values because of the aromatic ring current. |
| Γ b) Arene C-H chemical shift (δ) values are smaller than simple alkenes C-H chemical shift values because of the aromatic ring current. |
| C c) Arene C-H signals are always multiplets. |
| C d) Arene C-H signals are always singlets. |
| 4. Which of the following statements is wrong? |
| C a) UV absorption is attributable to electronic transitions. |
| UV spectra provide information about valence electrons. |
| C c) IR absorption is attributable to transitions between rotational energy levels of whole molecules. |
| NMR spectrometers use radiofrequency electromagnetic radiation. |
| Which is the correct order of increasing wave number of the stretching vibrations of (1) C-H |
| (alkane), (2) O-H (alcohol), (3) C=O (ketone), and (4) C≡C (alkyne)? |

| □ (3) < (4) < (2) < (1) □ (3) < (4) < (1) < (2) □ (4) < (3) < (1) < (2) □ (4) < (3) < (1) < (2) How many signals does the aldehyde (CH₃) CCH₂CHO have in ¹H NMR spectra? How many signals does the aldehyde (CH₃) CCH₂CHO have in ¹H NMR spectra? How many signals does the aldehyde (CH₃) CCH₂CHO have in ¹H NMR spectra? How many signals does the aldehyde (CH₃) CCH₂CHO have in ¹H NMR spectra? |
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| 6. How many signals does the aldenyde \(\begin{align*} \text{L} & \text{a) five H signals and four DC signals} \\ \text{L} & \text{b) three H signals and four DC signals} \\ \text{L} & \text{b) three H signals and four DC signals} \\ \text{L} & \text{b) three H signals and four DC signals} \\ \text{L} & \text{c) three H signals and four DC signals} \\ \text{L} & \text{c) three H signals and four DC signals} \\ \text{L} & \text{c) three H signals and four DC signals} \\ \text{L} & \text{c) three H signals and four DC signals} \\ \text{L} & \text{c) three H signals and four DC signals} \\ \text{L} & \text{c) three H signals and four DC signals} \\ \text{L} & \text{c) three H signals and four DC signals} \\ \text{L} & \text{c) three H signals and four DC signals} \\ \text{L} & \text{c) three H signals and four DC signals} \\ \text{L} & \text{c} & \ |
| C b) three 'H signals and four 'C signals C c) five 'H signals and four 'C signals C d) three 'H signals and six 'C signals C d) three 'H signals and six 'C signals 7. Which of hydrogens a-d in the following molecule gives a triplet signal in a normal 'H NMR spectrum? |
| 7. Which of hydrogens and in the NMR spectrum? |
| O CH ₃ -C-CH ₂ CH(OCH ₃) ₂ a b c d |
| C a) hydrogen a |
| D b) hydrogen b |
| C c) hydrogen c C d) hydrogen d 8. Which hydrogen of 1-chloropent-2-ene shows the largest chemical (downfield) shift in its |
| a Which hydrogen of 1-chloropent-2-ene shows the largest chemical (a |
| NMR spectrum? |
| C a) the H on C1 |
| □ b) the H on either C2 or C3 |
| C c) the H on C4 |
| D d) the H on C5 Which carbon of (a)-(d) of hex-3-en-2-one shows the largest (most downfield) chemical |
| shift in the NMR spectrum? |
| C a) C1 |
| C b) C2 |
| C c) C4 |
| C d) C6 |
| 10. Which of the following statements regarding IR spectroscopy is wrong? |
| (a) Infrared radiation is higher in energy than UV radiation. |
| D) Infrared spectra record the transmission of IR radiation. |
| C c) Molecular vibrations are due to periodic motions of atoms in molecules, and |

include bond stretching, torsional changes, and bond angle changes.

 \Box d) Infrared spectra give information about bonding features and functional groups in molecules.

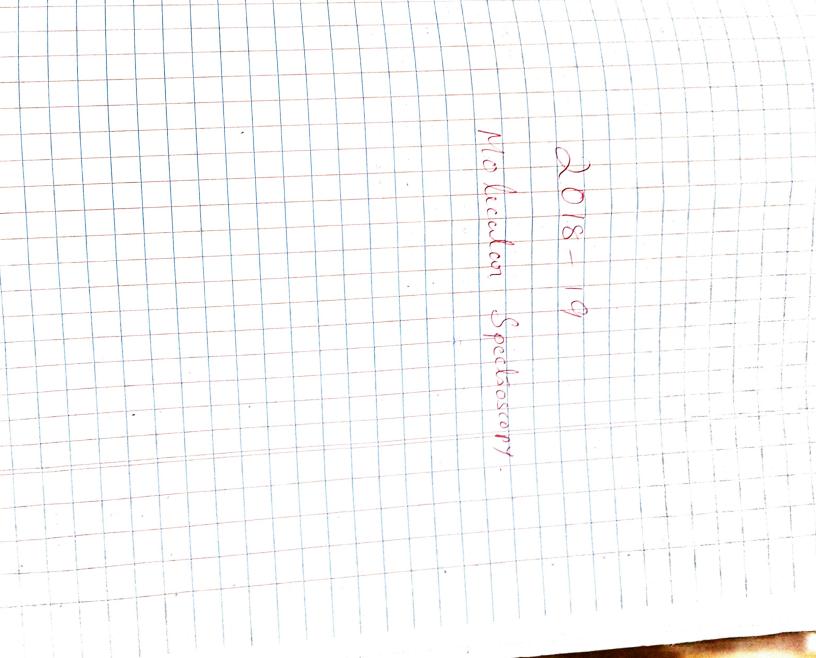
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2018-2019

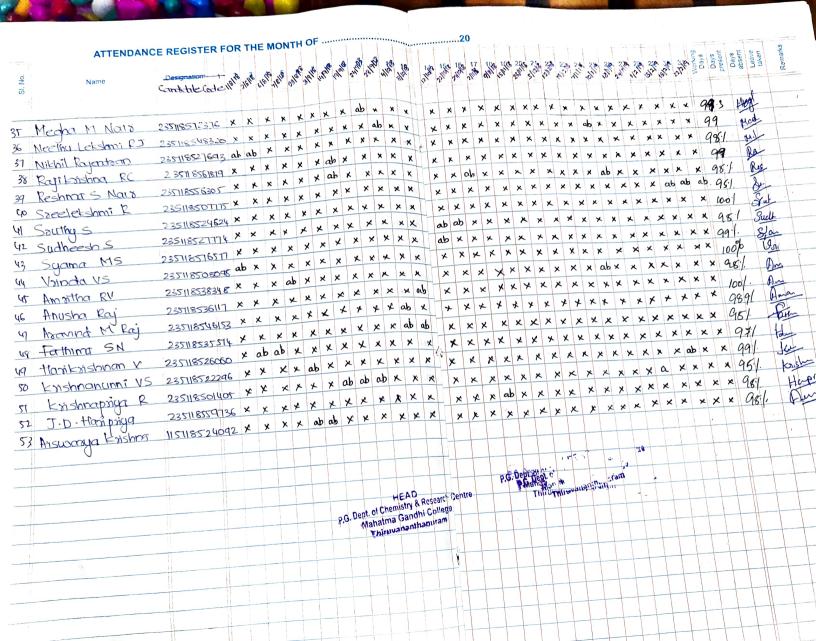
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| 1 | 6 | Anjana M.G. | 23 | 5118516210 | 9 |
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Thiruvananthapuram



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MAHATMA GANDHI COLLEGE

DEPARTMENT OF CHEMISTRY

CERTIFICATE OF COMPLETION

This is to certify that Keerthana R. S has successfully completed the Add On course entitled

" molecular spectroscopy"

Dr.Sikha T.S Head of the Department

31/03/2019

P.G. Dept. of Chemical Candhi Comp. P.G. Dept. of Chemical Candhi Comp. P.G. Nahatma Gandhi Canhapuran ya fairuvanantha puran ya fairuvanantha ya fairuvanantha puran ya fairuvanantha ya

A

Dr. Simi C.K. Course coordinator



MAHATMA GANDHI COLLEGE

DEPARTMENT OF CHEMISTRY

CERTIFICATE OF COMPLETION

This is to certify that Maneesha M has successfully completed the Add On course entitled

" MOLECULAR SPECTROSCOPY"



Dr.Sikha T.S Head of the Department

Course coordinator

Dr. Simi C.K.

31/03/2019



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SUMMARY REPORT OF ADD-ON COURSE - MOLECULAR SPECTROSCOPY (2018-2019)

course. A hand on trining was given to students and the practical session was very useful for the 01/08/2018 to 22/02/2019 for 30 hours. Theory and practical sessions were included in this 2018-2019, fifty three students were enrolled for the course. Classes were conducted from students. Add-on course on Molecular Spectroscopy was successfully completed. In the academic year

secured 60% marks and above in written examinations were honoured by certificates. After the completion of the course, students were assessed by written examination. Those who

